## GCE A LEVEL MARKING SCHEME

SUMMER 2017

A LEVEL (NEW)
PHYSICS - UNIT 3
1420U30-1

## INTRODUCTION

This marking scheme was used by WJEC for the 2017 examination. It was finalised after detailed discussion at examiners' conferences by all the examiners involved in the assessment. The conference was held shortly after the paper was taken so that reference could be made to the full range of candidates' responses, with photocopied scripts forming the basis of discussion. The aim of the conference was to ensure that the marking scheme was interpreted and applied in the same way by all examiners.

It is hoped that this information will be of assistance to centres but it is recognised at the same time that, without the benefit of participation in the examiners' conference, teachers may have different views on certain matters of detail or interpretation.

WJEC regrets that it cannot enter into any discussion or correspondence about this marking scheme.

## GENERAL INSTRUCTIONS

## Recording of marks

Examiners must mark in red ink.
One tick must equate to one mark (except for the extended response question).
Question totals should be written in the box at the end of the question.
Question totals should be entered onto the grid on the front cover and these should be added to give the script total for each candidate.
Marking rules
All work should be seen to have been marked.
Marking schemes will indicate when explicit working is deemed to be a necessary part of a correct answer.
Crossed out responses not replaced should be marked.
Credit will be given for correct and relevant alternative responses which are not recorded in the mark scheme.
Extended response question
A level of response mark scheme is used. Before applying the mark scheme please read through the whole answer from start to finish. Firstly, decide which level descriptor matches best with the candidate's response: remember that you should be considering the overall quality of the response. Then decide which mark to award within the level. Award the higher mark in the level if there is a good match with both the content statements and the communication statement.

Marking abbreviations
The following may be used in marking schemes or in the marking of scripts to indicate reasons for the marks awarded.

| cao | $=\quad$ correct answer only |
| :--- | :--- | :--- |
| ecf | $=\quad$ error carried forward |
| bod | $=\quad$ benefit of doubt |

bod = benefit of doubt

| Question |  |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | A01 | AO2 | AO3 | Total |  |  |
| 1 | (a) |  |  | Diagram showing source, absorber and detector or equivalent stated in words (1) <br> I. Take measurement with no source and no paper/aluminium [to measure the background radiation] or show awareness of background. <br> II. Measure count rate with no paper/aluminium. <br> ( 1 for both I and II) <br> III. Insert [a few sheets of] paper between the source and receiver and take measurement. If reduction from count in II then alpha particles present. (1) <br> IV. Insert [a few mm of] aluminium between the source and receiver and take measurement. If reduction in the count from that in III then beta particles are also present and if count is still above the background level, then gamma radiation is present this penetrates the aluminium. (1) | 4 |  |  | 4 |  | 4 |
|  | (b) | (i) | $\lambda=\frac{\ln 2}{T_{\frac{1}{2}}}=\frac{\ln 2}{(11.4 \times 24 \times 60 \times 60)}$ Substitution (1) $7.037 \times 10^{-7} \mathrm{~s}^{-1}$ unit (1) <br> Alternative solution: $\lambda=\frac{\ln 2}{T_{\frac{1}{2}}}=\frac{\ln 2}{11.4} \text { substitution (1) }=0.0608 \text { days }^{-1} \text { unit (1) }$ | 1 | 1 |  | 2 | 2 |  |
|  |  | (ii) | $\frac{57.0}{11.4}=5$ so 5 half lives (1) <br> Activity $=\frac{1}{2^{5}} \mathrm{~A}_{\mathrm{o}}=\frac{1}{32} \mathrm{~A}_{0} \quad$ (1) <br> Alternative solution: $\begin{aligned} & A=A_{0} e^{-\lambda t} ; t=5 T_{\frac{1}{2}} \text { and } \lambda=\frac{\ln 2}{T_{\frac{1}{2}}} ; \text { substitution } A=A_{0} e^{-5 \ln 2}(1) \\ & =A_{0} \frac{1}{2^{5}}=\frac{1}{32} A_{0}(1) \end{aligned}$ | 1 | 1 |  | 2 | 2 |  |


| Question | Marking details | Marks available |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total | Maths | Prac |
| (iii) | Let the initial number of particles be $N_{0}$, so number of particles remaining after 57 days $=\frac{1}{32} N_{0}$ i.e. $A \alpha N(1)$ <br> Decrease in the number of nuclei in 57 days $=\left(1-\frac{1}{32}\right) N_{0}$ i.e. decrease (1) <br> Percentage decrease $=\frac{\left(1-\frac{1}{32}\right) N_{0}}{N_{0}} \quad 100 \%=96.875 \%$ i.e. <br> percentage (1) <br> Answer only of 3\% award 2 marks only | 1 | $\begin{aligned} & 1 \\ & 1 \end{aligned}$ |  | 3 | 3 |  |
|  | Question 1 total | 7 | 4 | 0 | 11 | 7 | 4 |


| Question |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | AO1 | AO2 | AO3 | Total |  |  |
| 2 | (a) |  | The most stable nuclei (or reference to elements near to peak i.e. $\mathrm{Fe}, \mathrm{Ni} \mathrm{Ca}$ ) are therefore where the curve (or binding energy per nucleon) reaches its maximum. (1) <br> Nuclei of small atomic mass number (lhs of graph) can combine to produce species of larger atomic mass number, [hence larger binding energy per nucleon. Energy is released]. Fusion. (1) Nuclei of large atomic mass number (on rhs of graph) break down to produce species of smaller atomic mass number, [hence larger binding energy per nucleon. Energy is released]. Fission. (1) <br> Reference anywhere to there being energy released, when a reduction in mass occurs i.e. mass converted to energy. (1) | 4 |  |  | 4 |  |  |
|  | (b) | $\begin{aligned} & \text { Mass defect }=4(1.00728)+2(0.00055)(1) \\ & -4.00151=[0.02871 \mathrm{u}](1) \\ & \text { Energy: } 0.02871 \times 931=26.7[\mathrm{MeV}](1) \end{aligned}$ |  | 3 |  | 3 | 3 |  |
|  | (c) | Benefit: routine supply of energy (i.e. does not depend on weather) or jobs or no $\mathrm{CO}_{2}$ emission (1) <br> Issue: needs secure storage of radioactive waste (products) over a extended time period or reference to long half-lives or long build time or building extra transmission power lines (1) Reasoned conclusion (1) |  |  | 3 | 3 |  |  |
|  |  | Question 2 total | 4 | 3 | 3 | 10 | 3 | 0 |


| Question |  |  | Marking details |  |  | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | A01 | AO2 | AO3 | Total |  |  |
| 3 | (a) |  |  |  |  | $\Delta U$ increa <br> $Q$ heat <br> $W$ work <br> Reference to of 2 marks | ase (or change) in internal flowing into the system (1) done by the system (1) system at least once other | rgy of a system (1) <br> se award a maximum | 3 |  |  | 3 |  |  |
|  | (b) | (i) | Process <br> $A \rightarrow B$ <br> $B \rightarrow C$ <br> $C \rightarrow A$ | Description of process <br> Increase in pressure at <br> constant volume <br> Decrease in volume at <br> constant pressure (1) - <br> AO1 <br> [Linear] decrease in <br> pressure with increasing <br> volume (1) - AO1 | Work done on/by gas (if any) <br> No work done (1) AO2 <br> Work done on gas (1) - AO2 <br> Work done by gas (1) - AO2 | 2 | 3 |  | 5 |  |  |
|  |  | (ii) | Work done $\begin{align*} & =\frac{1}{2}(8-4) 10  \tag{1}\\ & =\frac{1}{2}(4)(1) 10 \tag{1} \end{align*}$ | on the gas is given by the " $\begin{aligned} & 0^{-3}(2-1) 10^{5} \\ & 0^{2}=200[\mathrm{~J}] \end{aligned}$ | " enclosed |  | 2 |  | 2 | 2 |  |
|  | (c) | (i) | Use of $P V=$ At A $P V=$ (or at C PV Check any p | $n R T$ and note $T$ constant (1) $\begin{aligned} & \left(1 \times 10^{5}\right)\left(8 \times 10^{-3}\right)=800 \\ & =\left(2 \times 10^{5}\right)\left(4 \times 10^{-3}\right)=8 \end{aligned}$ <br> point on curve e.g. | [J]) (1) | 1 | 1 |  | 3 | 3 |  |


| Question | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total |  |  |
| (ii) | Attempt at calculation of area - [accept small square count of between 170 and 280 i.e. $225 \pm \sim 25 \%$ ] (1) <br> If 225 i.e. 2.25 large squares, work difference $=$ $2 \Delta V \Delta p$ ( $\Delta V$ and $\Delta p$ for large square) <br> $=2.25\left(1 \times 10^{-3}\right)\left(0.2 \times 10^{5}\right)=45 \mathrm{~J}$ with uncertainty of $25 \%$, accept a value between 34 and 56 J provided method correct |  |  | 2 | 2 | 2 |  |
|  | Question 3 total | 6 | 7 | 2 | 15 | 7 | 0 |


| Question |  |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | A01 | AO2 | AO3 | Total |  |  |
| 4 | 4 (a) | a) |  | Increase in temperature - kinetic theory effects <br> Molecules move randomly <br> Collisions become more frequent [when heat supplied] <br> [No change in volume so heat flowing in] causes increase in $U /$ <br> kinetic energy <br> No work done [as constant volume] <br> $T$ increases with $U$ as temperature proportional to $U$, or equivalent <br> Newton's laws of motion <br> Momentum of molecules increase <br> Force on molecules = rate of change of momentum (during collision with wall) <br> Force on wall is equal and opposite to force on molecules Greater forces during the collisions <br> Increase in pressure <br> Molecules collide with walls exerting force on walls and / or pressure <br> Pressure increases with temperature <br> Pressure = force on walls per unit area <br> Mean pressure due to many collisions [and many molecules] <br> 5-6 marks <br> Comprehensive account including reference to increase in temperature, Newton's laws of motion and increase in pressure. There is a sustained line of reasoning which is coherent, relevant, substantiated and logically structured. <br> 3-4 marks <br> Comprehensive account including reference to 2 out of 3 of increase in temperature, Newton's laws of motion and increase in pressure or brief account of all 3 areas. <br> There is a line of reasoning which is partially coherent, largely relevant, supported by some evidence and with some structure. | 6 |  |  | 6 |  |  |


| Question |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total |  |  |
|  |  |  | 1-2 marks <br> Comprehensive account including reference to one of increase in temperature, Newton's laws of motion and increase in pressure or limited account of 2 areas. <br> There is a basic line of reasoning which is not coherent, largely irrelevant, supported by limited evidence and with very little structure. <br> 0 marks <br> No attempt made or no response worthy of credit. |  |  |  |  |  |  |
| (b) | (i) | $\begin{align*} & \text { rms }=\sqrt{\frac{400^{2}+425^{2}+450^{2}+550^{2}+625^{2}}{5}}  \tag{1}\\ & =497\left[\mathrm{~ms} \mathrm{~s}^{-1}\right] \text { (1) } \end{align*}$ |  | 2 |  | 2 | 2 |  |
|  | (ii) | The expected rms is explained by: $P V=n R T \quad \text { so } \quad P=\frac{n R T}{V}$ <br> $\rho=$ mass $/ V=\frac{n\left(M_{r} \times 10^{-3}\right)}{V}$ use of both equations by substitution (1) substitute these into: $P=\frac{1}{3} \rho \overline{c^{2}} ; \quad \sqrt{\overline{c^{2}}}=\sqrt{\frac{3 P}{\rho} ;}$ $\begin{aligned} & \sqrt{\overline{c^{2}}}=\sqrt{3 \frac{n R T}{V} \frac{V}{n\left(M_{r} \times 10^{-3}\right)}}=\sqrt{\frac{3 R T}{\left(M_{r} \times 10^{-3}\right)}} \\ & \sqrt{\overline{c^{2}}}=\sqrt{\frac{3(8.31)(293)}{\left(32 \times 10^{-3}\right)}}=478\left[\mathrm{~m} \mathrm{~s}^{-1}\right](1) \end{aligned}$ <br> Valid conclusion with data i.e. yes, the rms speed of the five molecules is slightly higher [about $4 \%$ above the expected rms of the gas] (1) ecf <br> Alternative solution: Use $m \overline{c^{2}}=3 k T$ $\begin{equation*} \sqrt{\overline{c^{2}}}=\sqrt{\frac{3 k T}{m}}=\sqrt{\frac{3\left(1.38 \times 10^{-23}\right)(293)}{(32)\left(1.66 \times 10^{-27}\right)}} \tag{1} \end{equation*}$ <br> with sensible substitution ( 1 ) $=478\left[\mathrm{~m} \mathrm{~s}^{-1}\right]$ |  | 1 | 1 <br> 1 | 3 | 3 |  |


| Question | Marking details | Marks available |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total | Maths | Prac |
| (iii) | Density does not change and <br> Pressure increases from the original pressure $p$ to $1.2 p$ (1) i.e. an increase of $20 \%$. <br> so $\begin{align*} & \left(\sqrt{\overline{c^{2}}}\right)_{\text {new }}=\sqrt{\frac{3(1.2 p)}{\rho}} \\ & \left(\sqrt{\overline{c^{2}}}\right)_{\text {new }}=\sqrt{1.2} \sqrt{\frac{3 p}{\rho}}=\sqrt{1.2}(478)=524 \mathrm{~m} \mathrm{~s}^{-1} \tag{1} \end{align*}$ |  | 2 |  | 2 | 2 |  |
|  | Question 4 total | 6 | 5 | 2 | 13 | 7 | 0 |



| Question |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total |  |  |
| (c) | (i) |  | $\begin{aligned} & \text { Centripetal force }=0.010 \frac{v^{2}}{R} \\ & \left(1 \text { for } \frac{m v^{2}}{R} ; 1 \text { if value inserted for } m\right) \end{aligned}$ | 1 | 1 |  | 2 | 1 | 2 |
|  | (ii) | Forces acting on mass Q: $0.090 g-\tau=0 \quad \tau$ :tension (1) So $\tau=0.090 \mathrm{~g}$. Substitution for $\tau$ into (c)(i) (1) $\begin{aligned} & 0.090 g=0.010 \frac{v^{2}}{R} \\ & v^{2}=\frac{0.090 g}{0.010} R \\ & v^{2}=9 g R \text { clear and convincing working (1) } \end{aligned}$ |  | 3 |  | 3 | 2 | 3 |




| Question |  |  |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  |  | A01 | AO2 | AO3 | Total |  |  |
| 6 | (a) | (i) |  | $T$ tension in the string and $m g$ weight of mass (gravitational force or gravity) | 1 |  |  | 1 |  |  |
|  |  | (ii) |  | $T$ does not have a component tangential to the arc (1) Component of $m g$ tangential to the arc is $m g \sin \theta$, (1) this is in the opposite direction to $s($ or $\theta)$ and so the negative sign (1) |  | 3 |  | 3 | 1 |  |
|  |  | (iii) |  | $\begin{aligned} \text { acceleration }= & \frac{-m g \sin \theta}{m}=-g \sin \theta \\ & =-g \theta(1)(\text { using the approximation) } \\ & \theta=\frac{s}{l} \text { or } \sin \theta=\frac{s}{l}(1)=-\frac{g s}{l} \end{aligned}$ |  | 2 |  | 2 | 2 |  |
|  |  | (iv) |  | Acceleration $\propto \theta$ (or $s$ ) measured [from a fixed position] (1) and opposite in direction (-ve) so SHM (1) |  |  | 2 | 2 |  |  |
|  | (b) | (i) | I | Substitution: $T=2 \pi \sqrt{\frac{l}{g}}=2 \pi \sqrt{\frac{1.2}{9.81}}$ (1) $=2.20$ [s] (1) | 1 | 1 |  | 2 | 2 |  |
|  |  |  | II | $f=\frac{1}{T}=0.45[5][\mathrm{H}] \mathrm{z} \mathrm{ecf} \mathrm{(1)}$ |  | 1 |  | 1 | 1 |  |
|  |  | (ii) |  | For maximum distance along the $\operatorname{arc} \theta_{\max }=0.067$, also $\sin \theta_{\text {max }}=0.067$ <br> As $\theta_{\text {max }}=\sin \theta_{\text {max }}$ (i.e. for the largest value of $\theta$ ) (1) then $\sin \theta$ is equal to $\theta$ for all $\theta$, and approximation holds for SHM. System oscillates with SHM (1) |  |  | 2 | 2 | 1 |  |
|  |  |  |  | Question 6 total | 2 | 7 | 4 | 13 | 7 | 0 |


| Question |  |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  | A01 | AO2 | AO3 | Total |  |  |
| 7 | (a) |  |  | Microwave laser or amplifier or equivalent (1) from water [molecules] or water clouds or steam (1) in gas disk around (supermassive) black hole (or quasar) (1) pumping or excitation provided by collisions (accept light) (1) (em radiation) propagates outwards (away from black hole) or reference to population inversion (1) <br> Treat as neutral reference to watermaser. <br> (Don't award the individual marks if later contradicted) | $\begin{aligned} & 1 \\ & 1 \\ & 1 \end{aligned}$ | $\begin{aligned} & 1 \\ & 1 \end{aligned}$ |  | 5 |  |  |
|  | (b) |  | Appropriate wavelength or energy or frequency chosen e.g. $400 \mathrm{~nm}-700 \mathrm{~nm}$ or $2-3 \mathrm{eV}$ or $4-8 \times 10^{14} \mathrm{~Hz}(1)$ <br> Valid method for obtaining ratio of frequencies, wavelength or energy (1) <br> Answer $35 \mathrm{~m}[\mathrm{~s}]-80 \mathrm{~m}[\mathrm{~s}]$ (1) | 1 | $\begin{aligned} & 1 \\ & 1 \end{aligned}$ |  | 3 | 2 |  |
|  | (c) |  | $\begin{aligned} & \frac{3300}{68.9}(1) \\ & 48[\mathrm{Mpc}](1) \end{aligned}$ |  | 2 |  | 2 | 2 |  |
|  | (d) |  | Measure the velocity using Doppler shift or use of Doppler equation (1) <br> at different times (1) <br> acceleration = rate of change of velocity (1) <br> Alternative: <br> Measure the velocity using Doppler shift (1) <br> Obtain $v_{\text {max }}$ (1) <br> $a=\frac{v^{2}}{r} \frac{\text { max }}{}$ or equivalent e.g. use $v$ and $r$ to calculate $a(1)$ |  |  | 3 | 3 |  |  |


| Question |  | Marking details | Marks available |  |  |  | Maths | Prac |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | A01 | AO2 | AO3 | Total |  |  |
| (e) | (i) |  | Use of $r=\frac{v^{2}}{a}(1)$ <br> Acceleration conversion i.e. /365/24/3600 (1) <br> Answer $=8.8 \times 10^{14}[\mathrm{~m}]$ or $8.8 \times 10^{11} \mathrm{k}[\mathrm{m}](1)$ |  |  | 3 | 3 | 3 |  |
|  | (ii) | Approximation used i.e. $D=\frac{r}{\theta}(1)$ <br> Answer $=7.8 \times 10^{14}[\mathrm{~m}](1)$ <br> Hence consistent (since overlap) i.e. valid conclusion based on calculations (1) <br> Comparing e.g. $1.53 \pm 0.15$ and $1.73 \pm 0.17$ but also accept combined error $=20 \%$ or $8.8 \times 10^{14}$ is less than $20 \%$ bigger than $7.8 \times 10^{14}(1)$ |  |  | 4 | 4 | 2 |  |
|  |  | Question 7 total | 4 | 6 | 10 | 20 | 9 | 0 |

A2 UNIT 3 - OSCILLATIONS and NUCLEI
SUMMARY OF MARKS ALLOCATED TO ASSESSMENT OBJECTIVES

| Question | A01 | AO2 | AO3 | TOTAL MARK | MATHS | PRAC |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 7 | 4 | 0 | 11 | 7 | 4 |
| 2 | 4 | 3 | 3 | 10 | 3 | 0 |
| 3 | 6 | 7 | 2 | 15 | 7 | 0 |
| 4 | 6 | 5 | 2 | 13 | 7 | 0 |
| 5 | 1 | 13 | 4 | 18 | 15 | 18 |
| 6 | 2 | 7 | 4 | 13 | 7 | 0 |
| 7 | 4 | 6 | 10 | 20 | 9 | 0 |
| TOTAL | 30 | 45 | 25 | 100 | 55 | 22 |

